

THEORETICAL EVALUATION OF THE CHANGE IN DETONATION CHARACTERISTICS OF ENERGETIC AMINES UPON CONVERSION INTO TRIAZENES AND PENTAZOLES

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Aromatic and heterocyclic amines form a huge pool of modern energetic materials, both molecular and salt-like [1]. Simultaneous presence of amino and nitro groups in an energetic molecule provide benefit in stability due to strong intramolecular hydrogen bonds, like in 1,3-diamino-2,4,6-trinitrobenzene (DATB) and 1,3,5-triamino-2,4,6-trinitrobenzene (TATB), which results in high h_{50} values equal 320 and 490 cm, respectively [2]. Additionally, such amines can be converted into diazoaminoarenes (triazenes) and pentazoles via formation of diazolum salt as an intermediate (Fig. 1) [3]:

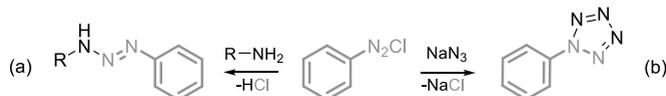
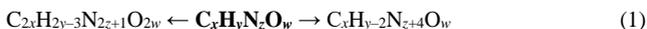


Fig. 1. Schematic presentation of the reactions studied in this work

According to our recently proposed method [4], these reactions can be schematically illustrated as the following (left – triazene, right – pentazole):



As a result, one can expect the following increase in density and heat of formation (Table 1).

Table 1. General expression for gain in density (Δd_c) and heat of formation ($\Delta\Delta H_f$)

Route	Δd_c	$\Delta\Delta H_f$
(a)	$\frac{1.020(a_1x + b_1y + c_1z + w)}{(a_2x + b_2y + c_2z + w)(a_2x + b_2y + c_2z + w - 0.091)}$	$-1.2845(-32x + 26y - 89z + 56w - 167)$
(b)	$\frac{2.398(a_3x + b_3y + c_3z - w)}{(a_2x + b_2y + c_2z + w)(a_2x + b_2y + c_2z + w + 4.386)}$	$1.2845(4N - 2H) = 524.1 \text{ kJ mol}^{-1}$

Herein, $a_1 = 1.457$, $b_1 = 0.418$, $c_1 = 1.255$, $a_2 = 1.643$, $b_2 = 0.512$, $c_2 = 1.355$, $a_3 = 2.201$, $b_3 = 1.423$ and $c_3 = 0.712$. Thus, one can easily calculate how much the detonation characteristics will increase for an arbitrary composition $C_xH_yN_zO_w$, which corresponds to an aromatic or heterocyclic amine. This is a convenient method for a crude estimation of the detonation properties without any quantum-chemical calculations. Of course, the latter are still needed for a precise estimation, but now these expensive methods can be applied for a much shortened list of potential candidates.

References

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